Structure of Sch 528647: A New Antitumor Antibiotic Related to Fumagillin

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(Received for publication July 16, 2001)

Fumagillin (1) was originally isolated from Aspergillus fumigatus¹) as an antiparasitic,²) and belongs to the sesquiterpene family of compounds. In later biological studies, fumagillin and its analogs were reported as highly potent angiogenesis inhibitors, which inhibit endothelial cell proliferation in vitro and tumor-induced angiogenesis in vivo. $^{3\sim5}$ Methionine aminopeptidase-2 (MetAP-2), the specific enzyme was recently discovered to be the molecular target of fumagillin.^{6,7)} The binding mode of fumagillin to the enzyme was further revealed by X-ray crystallographic analysis.⁸⁾ Furthermore, a semisynthetic analog of fumagillin, TNP-470 (AGM-1470), is currently in phase II clinical trials for the treatment of patients with a variety of cancers, including Kaposi's sarcoma, breast, cervical, lung and renal cancer.^{9~13)} In order to provide pure material for biological studies in our anti-cancer program, as well as to search for some new analogs, an isolation project of fumagillin and related compounds was launched. In the process of large-scale preparation of fumagillin, Sch 528657 (2), a novel metabolite closely related to fumagillin was discovered from Aspergillus fumigatus as shown in Fig. 1. Herewith, we wish to report on the isolation, structure elucidation and biological activity of 2.

The crude ethyl acetate extract from 100 liters fermentation broth was chromatographed on silica gel column eluting with hexane and ethyl acetate (1:1) to obtain the enriched mixture of 1 containing a small amount of 2 (~2%). The mixture was further purified by reversed phase HPLC under following conditions: YMC-ODS preparative column, S-10P, 120 Å, 250×50 mm with a guard column 50×50 mm; $5 \sim 50\%$ acetonitrile in NH₄OAc

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(10 mM, as is pH) with a linear gradient for 20 minutes, then 50~100% acetonitrile/NH₄OAc for 15 minutes, and followed by a 100% acetonitrile holding for 10 minutes; 50 ml/minute flow rate; UV detection at 390 nm. Pure **2** was obtained as a yellow solid (16 mg), m.p.=144~146°C, $[\alpha]_{D}^{25} - 12.18^{\circ}$ (c 0.1, MeOH).

The molecular weight of 2 was determined to be 442 Da based on LC-MS data that showed the protonated molecular ion at m/z 443 (M+H)⁺ and a small water adduct ion at m/z460 $(M+H_2O)^+$ in the positive atmospheric pressure chemical ionization (+APCI) mode. This was also confirmed by observation of m/z 441 (M-H)⁻ in the negative APCI mode. The molecular formula was deduced by elemental analysis as $C_{26}H_{34}O_6$ (Anal: C, 70.52; H, 7.58; calcd for C₂₆H₃₄O₆: C, 70.59; H, 7.69). The UV spectral data of 2 showed a typical polyene pattern with maximum absorptions at 240 (72), 320 (sh. 1012), 335 (1478), 358 (1338) nm. As listed in Table 1 and 2, the ¹H and ¹³C NMR spectra of **2** were strikingly similar to fumagillin. In the ${}^{13}C$ NMR spectrum of 2, the olefinic methylene and quaternary carbon signals at 110.83 and 146.78 ppm indicated a terminal double bond formation in comparison with the spiro-epoxide carbon resonances at 51.66 and 60.28 ppm in **1**. The ¹H NMR data of **2** were consistent with the ${}^{13}C$

Fig. 1. Structures of fumagillin and Sch 528647.



#	1	2	#	1	2
1a	2.56 d, J=5⁵	4.90 s	14	1.65 s	1.64 s
1b	2.97 d, J=5	4.71 s	15	1.17 s	1.27 s
3	1.11, 2.10 m,m	2.18, 2.29 m,m	16	3.37 s	3.33 s
4	1.90 m	1.57, 1.93 m,m	2'	6.02 d, J=15	6.01 d, J=15
5	5.67 m	5.54 m	3'	7.31 dd, J=12,15	7.28 dd, J=12,15
6	3.67 dd, J=3,10	3.40, dd, J=3,10	4'	6.65 t, J=12	6.64 t, J=12
7	1.98 d, J=10	2.26 d, J=10	5'	6.79, t, J=12	6.76 t, J=12
9	2.58 t, J=6	2.61 t, J=6	6'	6.82 t, J=12	6.81 t, J=12
10	2.22 m	2.28 m	7'	6.65 t, J=12	6.64 t, J=12
11	5.24 t, J=8	5.23 t, J=8	8'	7.39 dd, J=12,15	7.36 dd, J=12,15
13	1.72 s	1.70 s	9'	6.06 d, J=15	6.04 d, J=15

Table 1. 1 H NMR spectral data of 1 and 2^{a} .

a. Recorded at 400 MHz in acetone $-d_6$

b. Coupling constants in Hz

#	1	2	#	1	2
1	51.66 t	110.83 t	14	18.09 q	18.07 q
2	60.28 s	146.78 s	15	14.59 q	14.09 q
3	30.23 t	31.32 t	16	56.77 q	56.73 q
4	26.50 t	29.03 t	1'	168.25 s	168.95 br.s
5	67.75 d	68.25 d	2'	124.87 br.d	125.43 br.d
6	80.40 d	81.46 d	3'	144.23 d	144.06 d
7	49.16 d	53.16 d	4'	134.85 d	134.34 d
8	59.08 s	60.33 s	5'	139.83 d	139.69 d
9	60.94 d	61.45 d	6'	140.53 d	140.66 d
10	28.20 t	28.47 t	7'	134.99 d	135.12 d
11	120.40 d	120.45 d	8'	144.74 d	144.87 d
12	134.85 s	134.53 s	9,	123.79 d	123.62 d
13	25.92 q	25.90 q	10'	166.44 s	166.65 s

Table 2. 13 C NMR spectral data of 1 and 2^{a} .

a. Recorded at 100 MHz in acetone $-d_6$

b. Multiplicity was determined by APT and DEPT data.

Fig. 2. HMQC-TOCSY data of 2.







Fig. 3. Some important HMBC data of 2.



NMR data showing two vinyl singlets at 4.71 and 4.90 ppm instead of the two epoxide doublets at 2.56 and 2.97 ppm in **1**. Extensive 2D-NMR experiments were carried out in order to elucidate the structure of **2**. Three fragments A, B and C were further established based on HMQC-TOCSY data as shown in Fig. 2. The assignments of protons and carbons were completed by the analysis of HMBC data as depicted in Fig. 3. The relative stereochemistry of **2** was determined by NOESY experiments, which revealed its similarity to fumagillin as shown in Fig. 4.

Compound 2 exhibited fairly potent antitumor activity against melanoma cells (SKMEL-5 cells) in the soft agar assay with an $IC_{50}=250$ nM. However, compound 1 elicited better potency with an $IC_{50}=8$ nM. This result of a >31-fold potency decrease strongly suggested that the



spiro-epoxide functionality plays an important role for antitumor activity.

Soft agar assay: Dilutions of compounds are made in DMSO and applied to a mixture of agarose and complete media to form the bottom layer. The bottom layer is then placed into specific wells and left to solidify. SKMEL-5 cells are prepared at a concentration of 5000 cells/well. Additional dilutions of compounds are made and cells are added to this agarose/media mixture. This top mixture is placed into specific wells and left to solidify before placing to a 37°C incubator for 3 weeks. Cells are then strained with MTT (1 mg/ml) and scanned for colony counts. Total area is measured and calculated vs. compound concentration.

Acknowledgements

Authors are grateful to Mr. E. MAXWELL and Dr. B. DASMAHAPATRA for biological data, Mr. G. TORRACA and S. MITTELMAN for elemental analysis data, and Ms. D. SCOTT for preparing the manuscript.

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